

Effect of temperature on optical properties of CuInS₂ thin films

A. S. Meshram^{1*}, Y. D. Tembhurkar² and O. P. Chimankar³

^{1,2}Department of Physics, S. K. Porwal College Kamptee, Nagpur 441002, Maharashtra, India

³Department of Physics, R. T. M. Nagpur University, Nagpur 440033, Maharashtra, India

*Corresponding author: ameshram138@gmail.com

Available online at: www.isroset.org

Received: 18/Nov/2018, Accepted: 05/Dec/2018, Online: 31/Dec/2018

Abstract- Polycrystalline CuInS₂ thin films were grown by Spray pyrolysis method. This films have been deposited onto the glass substrate by varying substrate temperature from 275⁰C, at the interval of 25 to 350⁰C by using aqueous solution of Cupric chloride, Indium tri-chloride and thiourea of 0.02 M and Seems to be more important parameters affecting physical properties of the semiconductor. On account for the temperature dependent optical properties of polycrystalline CuInS₂ thin films, grain boundary states were operative in nature. The values of absorption coefficient, extinction coefficient, refractive index, and dielectric constant have been calculated from the optical measurements. The prepared films at different substrate temperatures were found to have high transmittance (76 %) with the lowest thickness approximately $t = 0.126 \mu\text{m}$ and at low transmittance (50 %) of the films with highest thickness nearly equal to $t = 0.2163 \mu\text{m}$. Due to temperature variation, the atoms are arrange regular in manner compound of CuInS₂ films are investigates. As the films are not doped intentionally, defect observed in intrinsic nature by Sulphur interstitials. The most significant results in present study, the temperature varied with thickness of the film can be used to modify the optical band gap, extinction coefficient, refractive index and real-imaginary part of dielectric constants of CuInS₂ thin films.

Keywords- Spray pyrolysis, CuInS₂ Ternary Semiconductor, Optical Constants

I. INTRODUCTION

In recent years, the development of nano-structured I-III-VI₂ ternary compound semiconducting materials in the form of thin films occupy a prominent place in basic research and solid state technology due to their expanding range of potential applications [1] in the diverse field such as photovoltaic cells, electronic components, fabrication of large area photodiode arrays, photoconductors, sensors, antireflection coatings, optical filters, surface acoustic wave devices, solar selective coatings and solar cells etc [2-4]. The conversion efficiency of this area devices based on other chalcopyrites like CuInTe₂ [5] CuInS₂ [6-8] is almost found higher values. CuInS₂ is one of the I-III-VI₂ chalcopyrite-type mixed crystal semiconductor [2]. So many researcher [8] were prepared CuInS₂ thin films by spray pyrolysis method and studied some physical properties. CuInS₂ as an absorber layer is very promising for thin film photovoltaic applications [4], which matches well with the solar spectrum, its high absorption coefficient of approximately 10^4 cm^{-1} and its direct band gap energy obtained from 1.32 – 2.45 eV [3-5] at near the red edge of the visible spectrum and high absorption coefficient [5-6] and its controllable conduction type. CuInS₂ is less toxic than selenide based compounds like CuInS_{2x}Se_{2(1-x)}}, CdZnSe₂ and CuInTe_{2(1-x)}}Se_{2x}. As the temperature of the glass substrate increases,

band gap (E_{g1} and E_{g2}) values, extinction coefficient (k), imaginary part of dielectric constant (ϵ_1) increased whereas real part of dielectric constant (ϵ_2), refractive index (n) decreases with wavelength (λ) from 350 – 1100 nm respectively. Refractive index and thickness of CuInS₂ and CdS films was obtained using an Ellipsometer by other researcher [9]. Different method have been used to produce CuInS₂ thin films, with including electrodeposition [10], evaporation [4, 6 and 11-12], vapor deposition [2] spray pyrolysis [9] and Sputtering technique [13]. In sulfurization process, an elemental sulfur vapor avoid highly toxic H₂S [14], developed Cu-rich CuInS₂ films showed an energy band gap of 1.38 eV at absorption coefficient of about 10^5 cm^{-1} [9] [12].

II. RELATED WORK

In this paper we focused on the effects of temperature on preparing thin films at different temperatures in air atmosphere and studied optical properties of CuInS₂ films by spray pyrolysis method. This method is one of the best technique to producing polycrystalline films from all chemical compound of solid solution which are form aqueous solution and this system not only inexpensive, easy deposition in moderate temperature but also developed films in large area. UV- 1800 Shimadzu spectrophotometer used to find optical absorption edge from the evaluated spectra

from transmittance curve by varied substrate temperature for the preparation of films. We investigate, temperature dependence of optical properties and other related optical constants characterization of CuInS_2 thin films by varied temperatures (250 to 375°C at the interval of temperature 25°C).

III. EXPERIMENTAL DETAILS

The deposition was carried out onto commercially available glass substrates of the size (7.5 × 2.5 × 0.1) cm³. The analytical reagent grade chemicals used were Cupric chloride (CuCl_2), Indium tri-chloride (InCl_3) and thiourea ($(\text{NH}_2)_2\text{CS}$) were prepared of 0.02 M. Solution of CuCl_2 , InCl_3 and $(\text{NH}_2)_2\text{CS}$ in appropriate volumes in order to attain Cu:In:Te in the ratio 1:1:3.2. Stoichiometric amounts of the elements of 98-99% purity CuCl_2 , InCl_3 , and $(\text{NH}_2)_2\text{CS}$ were used to prepare the initial chemical compound in aqueous solution mixed for the formation of CuInS_2 thin films. Excess sulphur required to reduce the deficiency of sulphur [15-17]. If ratio of Cu:In:Te used in the ratio 1:1:2, shows the tellurium deficiency [18]. The aqueous solution was then sprayed as a fine mist containing the reactant molecules on the preheated glass substrate of kept at various temperature. By Bernoulli's action solution spray will take place through this nozzle of glass sprayer when air flows through the carrier gas nozzle and a fine mist is produced at the barrier which is allowed to deposit on the pre heated glass substrate [19]. In order to find optimized condition for deposition of CuInS_2 thin films, the depositions were carried out by varying one of the parameters as substrate temperature and keeping the others at fixed value. The sprayed droplets on reaching the hot substrate undergo pyrolytic decomposition and form a single crystal, cluster, or crystallites of the product [20]. Films of thicknesses ranging from 0.14-0.22 μm were deposited at various temperature for optical measurement at a room temperature. The films thickness was measured by Michelson-interferometer. Samples of various thicknesses (0.14-0.22 μm) were obtained. Electrical resistivity calculated by Four-probe technique. The experimental set-up we used for our sprayed process is diagrammed in Figure 1, and has been details described in references [16-17 and 20].

IV. OPTICAL PROPERTIES

Transmittance (T) and Reflectance (R) spectra of the CuInS_2 films was recorded at room temperature using UV-1800 Shimadzu Spectrophotometer for the wavelength range of 350 to 1100 nm. The absorption coefficient (α), Energy band gap (E_g) refractive index (n), extinction coefficient (k) and Dielectric constant value in Real and Imaginary Parts (ϵ_1 and ϵ_2) were obtained by utilizing these spectrum [22]. In our experiment we used this spectrophotometer for measuring the transmittance and reflectance spectra of the $\text{Ge}_{30}\text{Se}_{70}$ films [23] also deposited on glass substrate at various

temperatures. The high transmittance is probably due to the existence of an interfacial layer with low refractive index between CuInS_2 solid material on films and by glass. The absorption coefficient of all of solution sprayed films is close to 10⁴ cm⁻¹. This change in absorption edge can be related to the film thickness and temperature of the substrate, this behavior indicates an increase in the disorder with decreasing thickness and may be attributed to the introduction of some defects which create localized states in the band-gap and therefore increase the band-gap. Similar results were also observed by other researchers [5, 24]. It is also known that the thickness becomes important in the thin films. If the semiconductor thickness is larger than the absorption length, most light will be absorbed; and if the diffusion length is larger than the film thickness, most photo generated carriers can be collected. Their efficiencies have reached higher than ~15% [11, 25]. The dominant feature of the energy dependence of the absorption coefficient (α) is the onsets of absorption near the region of inter-band transition from valence to conduction bands. Figure (2) shows the variation of transmittance versus wavelength plots of the CuInS_2 thin films deposited using different chemical compound for temperature 250 to 375°C at the interval of 25°C (spectra of reflectance not taken here). As expected transmittance of spray deposited films increases with changing value of 'Temperature' in above mention interval. It was observed that transmittance (above 70 %) was constant for higher wavelength and then suddenly decreased [9]. The onset of decrease of transmittance depends upon the concentration and represents the fundamental absorption edge [15, 16, 22 and 27], whereas thickness values increases upto optimum temperature. In the absorption measurements, we are concerned with the light intensity 'I' after traversal of a thickness (t) of material as compared with the incident intensity 'I₀', thereby defining the absorption coefficient (α), $\alpha = 1/t \log (I_0/I)$ (1)

The absorption coefficient (α) which is the decrement ratio of incident radiation relative to unit length in the direction of wave propagation inside the medium is relating with the absorbance (A).

It can be seen from Fig.2 that the absorption coefficient decreases with wavelength (increasing the film thickness) exponentially, then becomes constant and reaches the value 10⁴ cm⁻¹ in the visible region. This result is very important because we know that the spectral dependence of the absorption coefficient affects the solar conversion efficiency [28]. It is observed from this graph that absorption coefficient decreases with varies the temperature and it is also found that the film thickness increase [29]. The absorption coefficient (α), is related to the extinction coefficient (k), by the following formula, $\alpha = 4\pi k/\lambda$ (2)

Where, λ is the wavelength.

In the present works absorption coefficient (α), calculates optical band gap values and with help of many researchers [5, 28 and 30] put the empirical equation (Tauc relation) between the optical energy gap (E_g) and energy of the incident photon which is,

$$\alpha h\nu = A(h\nu - E_g)^n \quad \dots\dots (3)$$

Here, α is the absorption coefficient, h is the plank constant, ν is the frequency of light, A is the characteristics constant as well as a function of density of states near the conduction and valance band edges, E_g is the optical band gap energy (eV), n is the integer and has a values $1/2$, $3/2$, 2 and 3 depending on the material and the type of the optical transition corresponding to the allowed direct, allowed indirect, forbidden direct and forbidden indirect transition respectively [5,30 and 31].

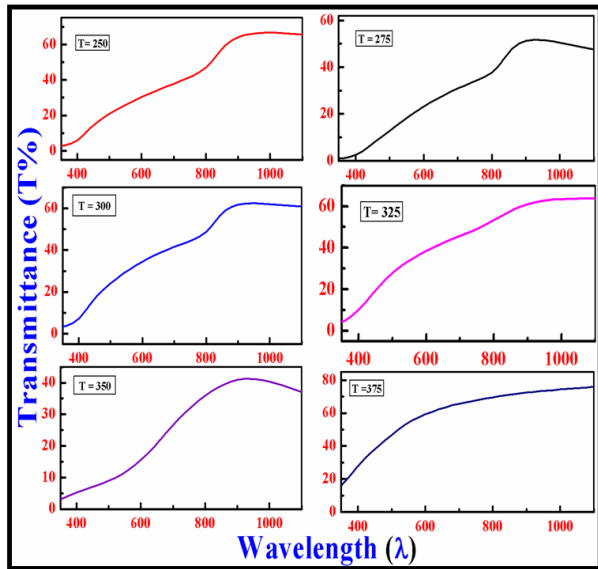


Fig. 2 Variation of Transmittance versus wavelength plots of CuInS₂ thin films deposited at 250^oC to 375^oC.

The optical band gap energy values for experimental thin films have been determined by using Tauc equation (3) which is used to find the type of the optical transition by plotting the square of the absorption coefficient (α) versus photon energy ($h\nu$) and extrapolating the straight line portion of the graph to the $h\nu$ axis (fig 3) [33]. This relation (3) is between absorption coefficient and optical band gap energy is expressed to calculate the band gap of the compounds by the following relationship. Since the plots are linear, the direct nature of the optical transition is confirmed [5, 33]. It is found that the relation for $n = 1/2$ yields linear dependence.

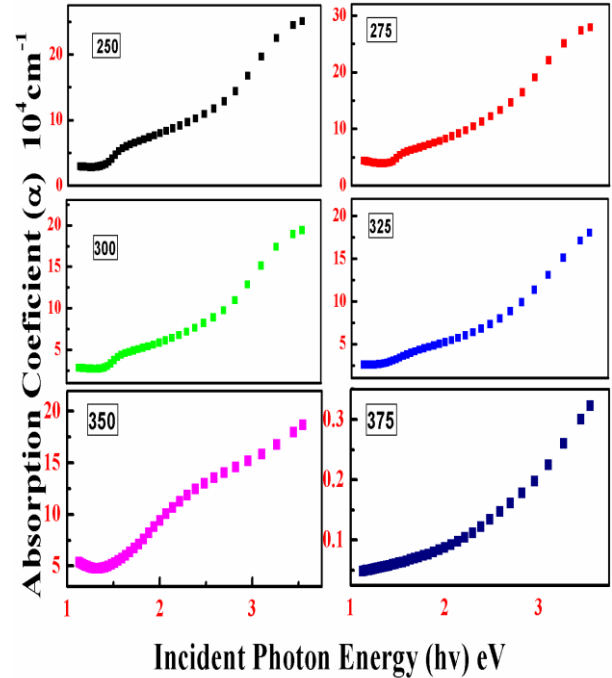


Fig. 3 Absorption coefficient versus incident photon energy $h\nu$ (eV) for various temperatures (250 to 375^oC)

The nature of optical band gap can be determined with the help of the relation $\alpha h\nu = A (h\nu - E_g)^n$, where absorption coefficient, A is a constant and n depends on the nature of the transitions. It is well known that the fundamental absorption for CuInS₂ was due to allowed direct transitions. Typical Fig. (4) Illustrates $(\alpha h\nu)^2$ as a function of photon energy ($h\nu$) for all thin films of as-deposited and CuInS₂ film with temperature 350^oC. The linear portion of each was extrapolated to $(\alpha h\nu)^2 = 0$ and meet energy ($h\nu$) axis given the values of band gap E_{g1} [34]. The absorption coefficient is calculated by using these values of band gap (E_{g1}) and slope A_1 of the curve, α_1 is calculated for $h\nu$. It can be seen from fig (6, 7) and tabulated in table. 1, that there is a shift in the band gap energy towards lower energies with the increase in the films thickness [12] and corresponds with temperature.

It was observed that calculated the value of α_1 is always less than observed absorption coefficient α for this range of wavelength. This can be only explained if we assume an additional absorption process [16, 35]. The absorption due to this additional process is denoted by α_2 , defined by relation [5],

$$\alpha_2 = \alpha_{exp} - \alpha_{1cal} \quad (4)$$

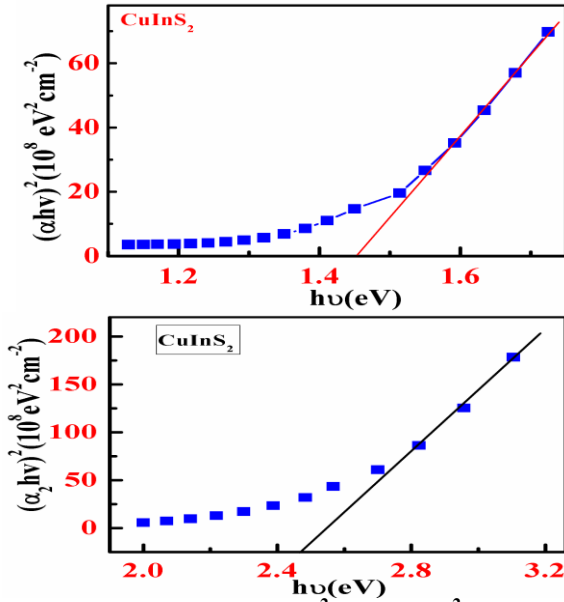


Fig. 4 Variation of $(\alpha hv)^2$ and $(\alpha_2 hv)^2$ with incident photon energy (hv) for typical $CuInS_2$ thin films deposited at $350^\circ C$.

Fig. 4 shows the graph $(\alpha_2 hv)^2$ versus $h\nu$ for the composition of typical $CuInS_2$ thin films as deposited at $350^\circ C$ temperature. It observed that the graph was linear. When it is extrapolated to $(\alpha_2 hv)^2 = 0$, the optical band gap (E_{g2}) is found to be 2.43 eV for $CuInS_2$ thin film prepared at $350^\circ C$ temperature. This also indicates the presence of second direct allowed transition.

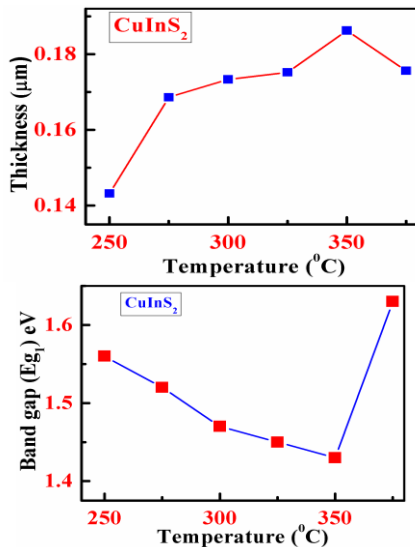


Fig. 5 Variation of thickness and optical band gap energy versus Temperatures

V. OPTICAL CONSTANTS

In general, optical constant [34] of any of the material is that any number of quantities characteristic of the optical behavior of a substance such as the refractive index,

absorption coefficient, or reflectivity for a specified wavelength, it is combination referred as called optical constant and have wide applications in designing different optical components and modeling optical coatings [36]. These values are evaluated from the transmittance and / or reflectance spectra.

Optical measurements can also be used to study lattice vibrations (phonons). This can be conveniently taken into account by defining a complex refractive index, $(n^* = n + ik)$. Here, the real part denoted by 'n' is the refractive index and it is also indicates the phase velocity of any materials, while the imaginary part k is called the extinction coefficient although k can also referred as mass attenuation coefficient and indicates the amount of attenuation when the electromagnetic wave propagates through the material. The real part of the refractive index n determines the propagation velocity (v and wavelength (λ) in the medium,

$$n = c / v \quad \dots (6)$$

The refractive index of materials varies with the wavelength (and frequency) of light. As the refractive index varies with wavelength, so will the refraction angle as light goes from one material to another.

For non-magnetic materials, we can take $\mu = 1$, the real and imaginary part of dielectric constant can be calculated by using the following equations,

$$\epsilon = \epsilon_1 + i\epsilon_2 = (n + ik)^2 \quad \dots (7)$$

$$(n - ik)^2 = \epsilon - i\epsilon_2 \quad \dots (8)$$

Where, $\epsilon_1 = n^2 - k^2 \quad \dots (9)$ And

$$\epsilon_2 = 2nk \quad \dots (10)$$

Where, n is the refractive index and k is the extinction coefficient and where we note that they are all frequency dependent.

The refractive index value can be calculated from the formula,

$$n = \left(\frac{4R}{(R-1)^2} - k^2 \right)^{1/2} - \frac{(R+1)}{(R-1)} \quad \dots (11)$$

And, $n = \frac{1 + \sqrt{R}}{1 - \sqrt{R}} \quad \dots (12)$

Where, R- is the reflectance, the extinction coefficient, which is related to the exponential decay of the wave as it passes through the medium.

Next, in order to complete the calculation of the optical constants, the extinction coefficient k is estimated from the values of absorption coefficient (k) and wavelength (λ),

$$k = \frac{\alpha \lambda}{4 \pi} \quad \dots (13)$$

Where λ is the wavelength of the incident radiation and absorption coefficient (α) is calculated by equation (1 and 2).

Plotting the graph in between k and incident photon energy ($h\nu$), therefore it illustrates the dependence of extinction coefficient (k) versus $h\nu$ for the studied thin film. The film thickness, energy band gap, extinction coefficient dielectric constant and refractive index of the films were calculated by the using the values of transmittance and reflectance data UV spectral graph, these values are listed in table 1.

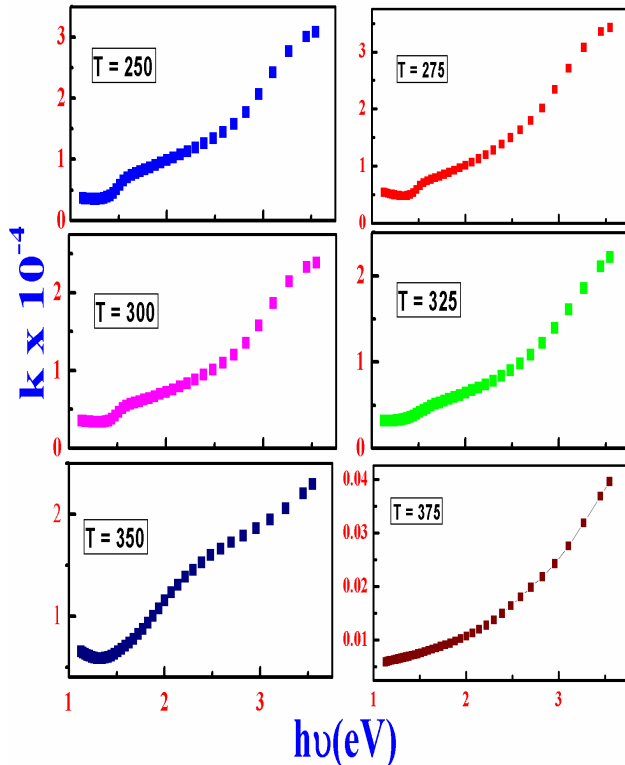


Fig. 6 Extinction coefficient versus incident photon energy (eV)

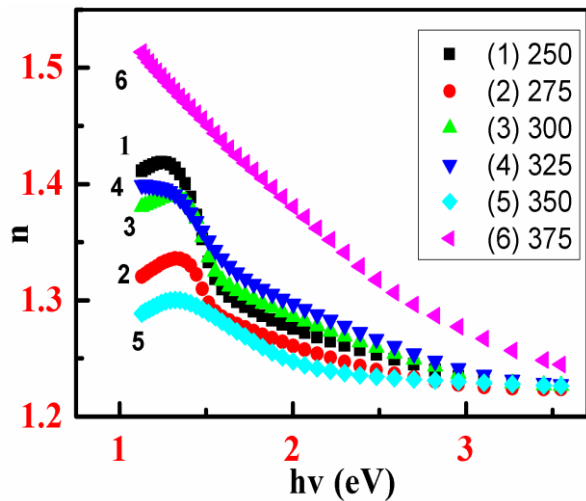


Fig. 7 Refractive index (n) versus incident photon energy ($h\nu$) (eV)

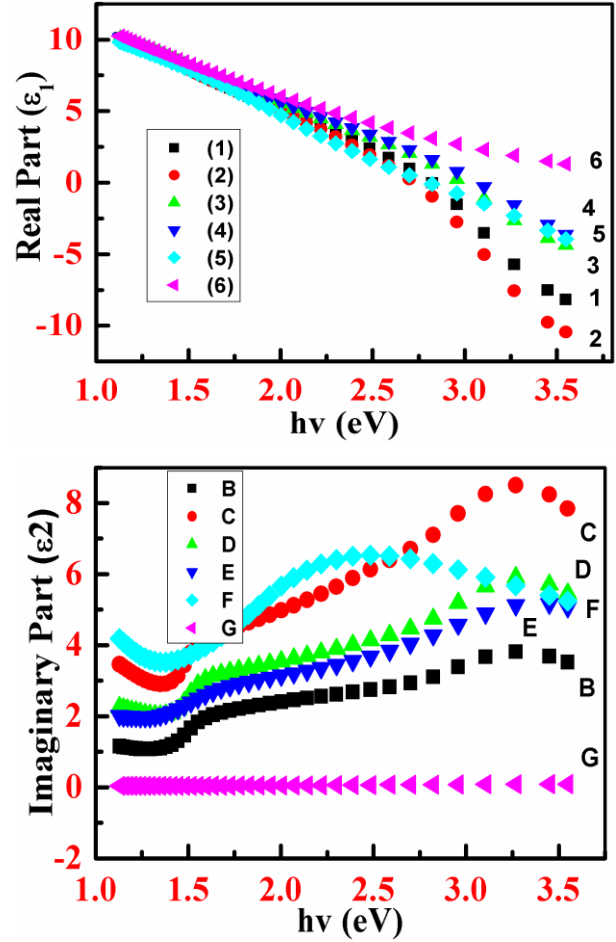


Fig. 8 Real (ϵ_1) and maginary (ϵ_2) Part of dielectric constant constant versus incident photon energy ($h\nu$) (eV)

Table (1) Tabulated values of Extinction Coefficient (k) and Refractive Index (n) real and imaginary part of Dielectric constant and other parameters with various temperatures of CuInS_2 thin films

| T (°C) | Thickn ess | Eg (eV) | Extincti on Coeffici ent (k) | Refracti ve Index | Dielectric constant | |
|--------|------------|---------|------------------------------|-------------------|---------------------|--------------|
| | | | | | ϵ_1 | ϵ_2 |
| 250 | 0.143 | 1.56 | 0.40-1.92 | 1.41-1.22 | 10.11-0.99 | 1.15-3.52 |
| 275 | 0.168 | 1.52 | 0.40-1.94 | 1.32-1.22 | 9.95-0.25 | 7.84-3.46 |
| 300 | 0.173 | 1.45 | 0.35-2.381 | 1.38-1.22 | 10.11-0.22 | 5.45-2.25 |
| 325 | 0.175 | 1.37 | 0.31-2.21 | 1.39-1.22 | 10.14-0.77 | 5.07-2.01 |

| | | | | | | |
|-----|-------|------|----------------|---------------|----------------|---------------|
| 350 | 0.186 | 1.43 | 0.65- 2.29 | 1.28- 1.22 | 9.81- 0.50 | 5.24- 4.18 |
| 375 | 0.175 | 1.65 | 0.005- 0.03 | 1.51- 1.24 | 10.24- 1.30 | 0.03- 0.09 |

VI. RESULTS AND DISCUSSION

The band gap values calculated from the Tauc plot for CuInS₂ thin films deposited at different substrate temperatures is shown in Fig. 3. The thickness was varied from 0.1432 to 0.1862 μm. Many researchers [12, 29 and 37] showed, as the thickness increases with band gap values are also changes and optimize in favorable parameters. The data shows that typical optical energy band gap and thickness values varies for all these thin films decreases with increasing substrate temperature (Fig. 6 and 7), indeed E_g for CuInS₂ increases from 1.37 to 1.65 eV and shows direct allowed transition [32]. The extrapolation of the straight line to $(\alpha h\nu)^2 = 0$ axis gives the value of the energy band gap [24, 38]. From figure (3) it is known that optical absorption coefficient and the transmittance of the thin films affects the E_{g1} value, these are 1.56, 1.52, 1.45, 1.37, 1.43 and 1.65 eV for the temperatures 250 to 375^oC respectively. The energy band values of CuInS₂ for these samples decrease with increase in substrate temperature upto 350^oC and further increase in substrate temperature [37], the band gap energy at 350^oC of the crystalline material confirms the stability of materials with nearly stoichiometry the variation of band gap energy with substrate temperature is shown in fig. 7. The increasing of optical energy band gap values with can be related with decreasing of defect states [23], no excess Cu₂S phase [31] and enhancement of crystalline structure [35] declared from XRD pattern [17]. On the other hand the increasing of band gap with the increasing of deposition temperature may be ascribed to increasing of transmittance hence which mean that the materials become more transparent (less absorbance). We noted a decrease of band gap value with the annealing temperature due the improvement of the crystallinity of the films.

The band gap for CuInS₂ was fairly in good agreement with result obtained produced by spray pyrolytically deposited films; the energy gap was 1.44 eV and 1.38- 1.5 eV respectively [31, 32 and 33]. This lower value of optical band gap may be due to the existence of tail states and traps. This is in agreement with the results published by other authors [39-40]. (E_{g1} = 1.43 – 1.5 eV). The calculation of the optical parameters of the structure of CuInS₂ thin films on glass substrate were obtained by mean transmittance values only using a method previously reported by Lontchi et al [4], was noted a decrease of band gap value with the annealing temperature due the improvement of the crystallinity of the films. Values was between 1.3 eV - 1.6

eV around the optimal band gap value for the un-doped and the Na-doped CuInS₂. The refractive index values was well agreed [4] values are reported results 2.6 and 2.7 for CuInS₂ and CdS films [9]. This low refractive index can be attributed to the singular ear structure, which could grow from first nucleation layer.

From Figure (4), it is observed that graph is a straight line, with extrapolated intercept on hv axis gives the values of band gap. In case of CuInS₂ the difference between two direct allowed transitions namely 1.43 and 2.48 eV is 1.06. Sridevi and Reddy [5] have reported thins value as 0.24 and have stated that this is spin-orbit splitting for flash evaporated CuInSe₂ thin films.

Using these equations (11, 12), easy to calculate the values of n and k respectively. The variations of refractive index n and extinction coefficient k with incident photon energy for CuInS₂ thin films of as-deposited and variation temperatures (275 to 375^oC) are shown in fig (7, 8). The extinction coefficient (k) values show a decreasing dependence on the photon energy and refractive index (n) values shows increases with incident photon energy as shown in figure (8). This coefficient characterizes absorption of the electromagnetic wave energy in the process of propagation of a wave through a material. It was also taken into account the scattering and luminescence effects. The values of refractive index n and the extinction coefficient (k) have been calculated at as deposited (350^oC), various temperatures (275 to 375^oC), and these values listed in Table (1). It is observed that the extinction coefficient decreases with varied temperature of the substrate of different films and it also corresponds thickness values varies [29] with temperature.

The real and imaginary parts of ϵ (ϵ_1 and ϵ_2) are related to the refraction index (n) and the extinction coefficient (k). Therefore, the complex dielectric function of a material and its variation with wavelength is used to interpret the transmission features and/or reflection spectra of semiconductor films. Real and imaginary parts of dielectric constant (ϵ_1 and ϵ_2) of CuInS₂ thin films deposited at various temperatures with different thickness were calculated.

Mahir N. Al-Jabery et al [39] were grown thin films. They are also extinction coefficient (k), refractive index (n) and the real and imaginary dielectric constants (ϵ_1 , ϵ_2) have been investigated. It may be also suggested that there small defects/ imperfect or non-stoichiometric results were observed at lower temperature [41]. So that temperature is important parameter for better results of films.

V. CONCLUSION AND FUTURE SCOPE

In Summary, we have prepared CuInS₂ thin films successfully by a simple and inexpensive spray pyrolysis

deposition method. The optical properties of CuInS_2 films were investigated in present works. We show that due to the increased temperature, the band gap and structure of the polycrystalline films get rearranged. The characterization of the CuInS_2 layer on glass substrates gave thin film with band gap value between 1.43 and 2.48 eV and absorption coefficient in the range of 10^4 cm^{-1} with a direct optical transition and it has been suggested that the reported results of thin films could be a suitable candidate for using as an absorber in photovoltaic applications. The other properties appear to be new to the improved material. The values of absorption coefficient, extinction coefficient, refractive index and dielectric constant were found to be in conformity with previous works.

ACKNOWLEDGEMENTS

Authors would like to express his thanks to Department of Physics, R.T.M Nagpur University Nagpur and Principal, S. K. Porwal College Kamptee for providing research facilities in the respective laboratories.

REFERENCES

- [1] David O. Scanlon and Graeme W. Watson, "Stability, geometry, and electronic structure of an alternative I-III-VI₂ material, CuScS_2 : A hybrid density functional theory analysis" Applied Physics Letter, Vol. 97, pp.131904-131904-3, 2010.
- [2] A. J. Nelson, C. R. Schwerdtfeger, G. C. Herdt, D. King, M. Contreras, K. Ramanathan and W. L. O'Brien, "X-ray photoemission analysis of chemically treated I-III-VI semiconductor surfaces", Journal Vacuum Science Technology A, Vol. 15, Issue 4, pp.2058-2062, 1997.
- [3] S. M. Pawar, B.S. Pawar, J.H. Kim, Oh-Shim Joo, C.D. Lokhande, "Recent status of chemical bath deposited metal chalcogenide and metal oxide thin films", Current Applied Physics Review, Vol.11, 11.117-161, 2011.
- [4] J. Lontchi, B. Khalfallah, M. Abaab, "Thermal Evaporated Undoped and Na-doped CuInS_2 with Copper Contact for Photovoltaic Applications", International Journal of renewable energy Research, Vol. 6 (2), pp.520-526, 2016.
- [5] D. Sridevi and K. V. Reddy, "Electrical conductivity and optical absorption in flash evaporated CuInTe_2 thin films", Thin Solid Films, Vol.141, 157-164, 1986.
- [6] A. Amara, W. Rezaiki, A. Ferdi, A. Hendaoui, A. Drici, M. Guerioune, J.C. Berne'de, M. Morsli, "Electrical and optical characterization of CuInS_2 crystals and polycrystalline co-evaporated thin films", Solar Energy Materials & Solar Cells, Vol. 91, pp.1916-1921, 2007.
- [7] J. E. Jaffe and A. Zunger, "Electronic structure of the ternary chalcopyrite semiconductors CuAlS_2 , CuGaS_2 , CuInS_2 , CuAlSe_2 , CuGaSe_2 , and CuInSe_2 ", Physical Review B, Vol. 28, pp.5822-5847, 1983.
- [8] K. M. A. Hussain, J. Podder, D. K. Saha and M. Ichimura, "Structural, electrical and optical characterization of CuInS_2 thin films deposited by spray pyrolysis", Indian Journal of Pure and Applied Physics, Vol. 50, pp.117-122, 2012.
- [9] P. M. Parameshwari, S. Bhat, K. Gopalakrishna Naik, "Structural, electrical and optical studies on spray deposited Cadmium Sulphide and Copper Indium Disulphide thin films", Achieve of Physics Research, Vol. 3(6), pp. 441-451, 2012.
- [10] X. H. Xu, F. Wang, J. J. Liu, K. C. Park, M. Fujishige, "A novel one-step electrodeposition to prepare single-phase CuInS_2 thin films for solar cells", Solar Energy Materials & Solar Cells, Vol. 95, pp. 791-796, 2011.
- [11] D. Braunger, D. Hariskos, T. Walter and H. W. Schock, "An 11.4% efficient polycrystalline thin film solar cell on CuInS_2 with a Cd-free buffer layer", Solar Energy Materials and Solar Cells, Vol. 40, pp. 97-102, 1996.
- [12] M. Ben Rabeah, N. Khedmi, M. A. Fodhaa, M. Kanzari, "The Effect of Thickness on Optical Band Gap and N-type Conductivity of CuInS_2 Thin Films Annealed in Air Atmosphere", Energy Procedia, Vol. 44, 52 - 60, 2014.
- [13] K. Ellmer, J. Hinze, J. Klaer, "Copper indium disulfide solar cell absorbers prepared in a one step process by reactive magnetron sputtering from copper and indium targets", Thin Solid Films, Vol. 413, pp.92-97, 2002.
- [14] YAN You-hua, LIU Ying-chun, FANG Ling, ZHU Jing-sen, ZHAO Hai-hua, LI De-ren, LU Zhi-chao, ZHOU Shao-xiong, "Characterization of CuInS_2 thin films prepared by sulfurization of Cu-In precursor", Transactions of Nonferrous Metals Society of China, Vol.18, pp. 1083-1088, 2008.
- [15] D. Sridevi and K. V. Reddy, "Preparation and characterization of $\text{CuInSe}_{2(1-x)}\text{Te}_{2x}$ solid solutions", Material Research Bulletin, Vol. 20, pp.929-934, 1985.
- [16] D. Sridevi J. J. B. Prasad and K. V. Reddy, "Preparation and characterization of flash-evaporated CuInSe_2 thin films", Material Research Bulletin. Vol. 8, pp.319-324, 1986.
- [17] D. Sridevi and K. V. Reddy, "Electrical conductivity optical absorption in flash-evaporated of CuInTe_2 thin films", Thin Solid Films, Vol. 141, pp.157-164, 1986.
- [18] G. Marin, S. M. Wasim, G. Sánchez Pèrez, P. Bocaranda and A. E. Mora, "Compositional, Structural, Optical and Electrical Characterization of CuInTe_2 Grown by the Tellurization of Stoichiometric Cu and In in the Liquid Phase", Journal of Electronic Materials, Vol. 27, No. 12, pp. 1351-1357, 1998.
- [19] N. Suriyanarayanan and C. Mahendran, "EDAX, SEM, Photoluminescence and Electrical properties of Zn doped polycrystalline CuInS_2 Thin films by spray pyrolysis", Archives of Physics Research, Vol. 3, pp. 54-59, 2012.
- [20] A. S. Meshram, Y. D. Tembhurkar, O. P. Chimankar, Structural, optical and electrical properties of CuInTe_2 thin films prepared by spray pyrolysis, International Journal of Advance Research in Science and Engineering, 6 (9) September (2017) 1735-1745.
- [21] P. Christian, A. M. Coclite, "Thermal studies on proton conductive copolymer thin films based on perfluoroacrylates synthesized by initiated Chemical Vapor Deposition", Thin Solid Films, Vol. 635, pp. 3-8, 2017.
- [22] S. Roy, P. Guha, S. Chaudhuri, A.K. Pal, " CuInTe_2 thin films synthesized by graphite box annealing In/Cu/Te stacked elemental layers", Vacuum, Vol. 65, pp. 27-37, 2002.
- [23] A. S. Solieman, M. M. Hafiz, Abdel-hamid A. Abu-Sehlyc, Abdelnaser A. Alfaqeer, "Dependence of optical properties on the thickness of amorphous $\text{Ge}_{30}\text{Se}_{70}$ thin films", Journal of Taibah University for Science, Vol. 8, pp. 282-288, 2014.
- [24] T. Mise, T. Nakada, "Low temperature growth and properties of Cu-In-Te based thin films for narrow band gap solar cells", Thin Solid Films, Vol. 518, pp. 5604-5609, 2010.
- [25] H. Goto, Y. Hashimoto, K. Ito, "Efficient thin film solar cell consisting of TCO/CdSCuInS₂/CuGaS₂ Structure", Thin Solid Films, Vol. 451-452, pp. 552-555, 2004.
- [26] M. Boustani, K. El Assali, T. Bekkay and A. Khiara "Structural and Optical properties of CuInTe_2 films prepared by thermal vacuum evaporation from a single source", Semicond. Sci. Technol., Vol.12, pp. 369-376, 1997.
- [27] R. H. Bari, L. A. Patil and P. P. Patil, "Structural, optical and electrical properties of chemically deposited nonstoichiometric

- copper indium diselenide films*", Bulletin of Material Science, Vol. 29, No. 5, pp. 529-534, 2006.
- [28] V. V. Kindyak, A. S. Kindyak, V. F. Gremenok, I. V. Bodnar, Y.V. Rud, G.A. Madvedkin, "Optical properties of laser-evaporated CuGaSe₂ films near and above the fundamental absorption edge", Thin Solid Films, Vol. 250, pp. 33-66, 1994.
- [29] S. Debnath, M.R. Islam and M.S. Khan, "Optical properties of CeO₂ thin films", Bulletin of Material Science, Vol. 30, pp.315-319, 2007.
- [30] T. Hurma, "Characterization of zincblende-CuInS₂ nanostructured films: The XRD, Raman, FT-IR and UV-Vis spectroscopical investigations", Indian Journal of Pure and Applied Physics, vol. 54, pp.797-801, 2016.
- [31] Y. Vahishad, R. Ghasemzadeh, A. Irajizad and S. M. Mirkazemi, "Synthesis and Characterization of Copper Indium Sulfide Chalcopyrite Structure with Hot Injection Method", Journal of Nanostructures, Vol. 3, pp.145-154, 2013.
- [32] A. Kotbi, B. Hartiti, A. Ridah and P. Thevenin, "Characteristics of CuInS₂ thin films synthesized by chemical spray pyrolysis," Optical and Quantum Electronics, Vol. 48 pp. 75 (1-9), 2016.
- [33] M. Ben Rabeh, M. Kanzari and B. Rezig, "Role of oxygen in enhancing N-type conductivity of CuInS₂ thin films", Thin Solid Films, Vol. 515, pp. 5943-5948, 2007.
- [34] A. Essahlaoui, H. Essaoudi, A. Hallaoui, M. Bouhadda, A. Labzour, A. Housni, "Calculation of the thickness and optical constants of lead titanate thin films grown on MgO from their transmission spectra", Journal of Material Environmental Science, Vol. 9 pp. 228-234, 2018.
- [35] M. Dhanam, R. Balasundarprabhu, S. Jaykumar, P. Gopalakrishnam, M. D. Kannam, "Preparation and study of structural and optical properties of chemical bath deposited copper indium diselenide thin films", Physics Status solidi (a), Vol.191,pp. 149-160, 2002.
- [36] S. K. Al-Ani, "Methods of Determining the Refractive Index of Thin Solid Films", Iraqi Journal of Applied Physics, Vol. 4, pp. 17-23, 2008.
- [37] N. Khedmi, M. Ben Rabeh, M. Kanzari, "Thickness Dependent Structural and Optical Properties of Vacuum Evaporated CuInS₂ Thin Films", Energy Procedia, Vol. 44, pp. 61 – 68, 2014
- [38] M. Lakhe and N. B. Chaure, "Low-temperature Heat Treatment (80°C) Effect on the Electrochemically Synthesized CuInTe₂ Thin Films for Energy Harvesting Applications", J. Material Science and Engineering, Vol. 4, Issue 6, pp.1-6, 2015
- [39] Mahir N. Al-Jabery, Qayes A. Abbas, Hamid S. Al-Jumaili, "Study the optical properties of CuInS₂ non stoichiometric thin films prepared by chemical spray pyrolysis method", Iraqi Journal of Physics, Vol.10, pp.70-75, 2012.
- [40] A. Bouloufa, Messous, M. V. Yakushev, R. D. Tomlinson, and A. Zegadi, "Optical Properties of CuInTe₂ Single Crystals by Photoacoustic Spectroscopy", Journal of Electron Devices, Vol. 2, pp. 34-39, 2003.
- [41] A. Rockett, "Fundamental Studies of the Effect of Crystal Defects on CuInSe₂/CdS Heterojunction Behavior", National Renewable Energy Laboratory (NREL) report 1998.

AUTHORS PROFILE

Mr. A. S. Meshram pursued M.Sc (Physics), doing Ph. D under the guidance of Dr. Y. D. Tembhurkar, Associate professor, S. K. Porwal College Kamptee, Nagpur, India and Dr. O. P. Chimankar, Associate Professor, Department of Physics, R. T. M. Nagpur University Nagpur, India. He is currently working as an Assistant professor of Physics, S. K. Porwal College Kamptee, Nagpur, India. His main research work focuses on thin films by Spray pyrolysis. He has 3 years of teaching and research experience in Junior and UG level.

Dr. Y. D. Tembhurkar has completed his Ph.D degree in the year 1992 from Institute of Science, Nagpur affiliated to R. T. M. Nagpur University, Nagpur, India. He has published more than 50 papers in international peer reviewed journal and attended 50 international conferences. Presently she is working as a Head and Associate professor, Department of Physics, S. K. Porwal College Kamptee, Nagpur, India. His main research work focuses on crystal growth, thin films and spray pyrolysis and characterization. He has 29 years of teaching and research experience in UG level.

Dr. O. P. Chimankar has completed his Ph.D degree, Nagpur affiliated to R. T. M. Nagpur University, Nagpur, India. Associate Professor, Department of Physics, R. T. M. Nagpur University Nagpur, India. He is currently working as Associate Professor in Department of Physics, R. T. M. Nagpur University, Nagpur, India. He has published more than 50 research papers in reputed international journals including Thomson Reuters (SCI & Web of Science) and conferences and it's also available online. He has more than 15 years of teaching experience and research experience.