

**International Journal of Scientific Research in \_ Research Paper . Physics and Applied Sciences** Vol.7, Issue.3, pp.182-189, June (2019) **E-ISSN:** 2348-3423 *DOI: https://doi.org/10.26438/ijsrpas/v7i3.182189*

# **Effect of post-annealing temperature on linear and non-linear optical properties of sol-gel spin coated CdS thin films**

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Received: 22/May/2019, Accepted: 12/Jun/2019, Online: 30/Jun/2019

*Abstract***—**Cadmium sulfide thin films were prepared by sol-gel spin coating method on glass substrates and its effect of postannealing temperature on linear and non-linear optical properties were investigated. X-ray diffraction results showed that the grown CdS thin films exhibit cubic phase with (1 1 1) as a preferential orientation. The surface morphology of CdS thin films was characterized by using atomic force microscopy. The electrical resistivity of CdS thin films measured from the four-probe method decreased from  $0.74 \times 10^4$  to  $0.12 \times 10^4$  Q.cm with the increase of post-annealing temperature from 150 to 300°C. The optical band gap of CdS thin films is found to be 2.42 eV, whereas its band gap decreased from 2.35 to 2.21 eV with the increase of post-annealing temperature from 150 to 300°C. The normal dispersion of refractive index for CdS thin films was described using Wemple-DiDomenico single-oscillator model. The optical dispersion parameters for CdS thin films were reported in this study. The Verdet constant is calculated based on the refractive index dispersion study. The first, third order nonlinear optical susceptibility  $(\chi^{(1)}, \chi^{(3)})$  and nonlinear refractive index n<sup>(2)</sup> were also determined.

*Keywords***—** Sol-gel spin coating; CdS thin films; X-ray diffraction; Atomic force microscopy; Optical properties.

# **I. INTRODUCTION**

Cadmium sulfide (CdS) had a wide applications in electrooptic devices such as photosensors, optical wave-guides, non-linear integrated optical devices, transducers, photoconducting cells, non-linear integrated optical devices, window layer in  $Cu(In,Ga)Se<sub>2</sub>$  (CIGS), CdTe heterojunction solar cells, field effect transistors, light emitting diodes, gas sensors and diluted magnetic semiconductors in spintronic devices [1-5]. CdS belongs to II-VI group semiconducting material having a direct band gap of 2.42 eV (in cubic phase) and 2.57 eV (in hexagonal phase) at room temperature. CdS exhibit in two forms namely α-CdS (hexagonal wurtzite, space group:  $P6<sub>3</sub>mc$ ) and  $\beta$ -CdS (cubic, zinc blende crystal structure, space group: F-43m). β-CdS forms at low temperature with a metastable phase whereas α-CdS forms at high temperatures with a stable phase; however, annealing at high temperatures can transform β-CdS to  $α$ -CdS [6].

There are different chemical and physical methods to prepare CdS thin films which include spray pyrolysis [7], metal organic chemical vapor deposition (MOCVD) [8], layer adsorption and reaction (SILAR) [9], sol-gel route [10], chemical bath deposition (CBD) [11], thermal evaporation [12], sputtering [13] and pulsed laser deposition (PLD) [14].

Among these methods, the sol-gel spin coating method is a low cost, simple and able to produce uniform films with good adherence, provides a high specific surface area, narrow pore size, better microstructural control of particles and uniform particle distribution [15,16].

Here, we report the structural, transport and optical properties of CdS thin films. Studies on the nonlinear optical properties of sol-gel spin coated CdS thin films are rare to found in the literature. The main objective of the present work is to study the effect of post-annealing temperature on linear and non-linear optical properties of CdS thin films.

# **II. RELATED WORK**

In the present work, we reported the detailed information on optical constants, optical dispersion parameters and nonlinear optical properties of CdS thin films which are of much importance for its applications in integrated optic devices such as modulators, filters, switches, etc.

# **III. METHODOLOGY**

CdS thin films were prepared by mixing 0.6 mL of polyethylene glycol (PEG 200) with 8.9 mL of ethanol and 0.5 mL of acetic acid under stirring continuously for 90 minutes. Cadmium nitrate  $(Cd(NO<sub>3</sub>)<sub>2</sub>.4H<sub>2</sub>O)$  and thiourea  $(SC(NH<sub>2</sub>)<sub>2</sub>)$  (chemicals of analytical grade from Sd Fine Chem. Pvt. Ltd., Mumbai, India) were used as precursors for cadmium and sulfur source. These precursors were slowly dissolved in ethanol and stirred for 90 minutes. The asprepared solution was slowly added to polyethylene glycol (PEG) sol with stirring for 6 hours in order to obtain the sol, which is used for preparing CdS thin films. The spin coating method was used to deposit CdS thin films on glass substrates using the above sol with a rotating spin speed of 1000 rpm for 45 seconds. The glass substrates were cleaned thoroughly with detergent laboline and then rinsed with double distilled water. Then the substrates were kept in hot chromic acid for 12 hours. After chromic acid treatment, the substrates were cleaned ultrasonically in double distilled water and acetone for 10 min. Finally, the substrates were dried under an infra-red (IR) lamp. These glass substrates were used for the preparation of CdS thin films. CdS thin films were prepared on glass substrates by sol-gel spin coating method at room temperature and post-annealed the films up to 300 $\degree$ C for 1 hour.

Structural properties of CdS thin films were analyzed by glancing angle X-ray diffraction (GAXRD) using a Rigaku X-ray diffractometer with CuKα radiation  $(λ=0.154$  nm) source operated at 40 kV and 40 mA. X-ray diffraction measurements were recorded in the 2θ range of 20°-90° at a glancing angle of  $2^{\circ}$  with a scan speed of  $1^{\circ}$  min<sup>-1</sup>. The surface morphology of the CdS thin films is studied by using atomic force microscope (AFM; Model: Park NX20). The thickness of the CdS films is measured by Dektak surface profilometer. The electrical measurements of CdS thin films were measured by the four-probe method and Hall effect setup. The optical transmittance spectra of the CdS thin films were recorded by using UV-Vis-NIR Spectrometer (Shimadzu MPC3600) in the wavelength range of 300-2500 nm.

#### **IV. RESULTS AND DISCUSSION**

## **3.1 Structural analysis**

X-ray diffraction (XRD) patterns of CdS thin films prepared by sol-gel spin coating method on glass substrates at room temperature (RT) and post-annealed in the vacuum from 150 to 300°C are shown in Figure 1. The diffraction peaks observed at  $2\theta = 26.46^{\circ}$ , 44.2°, and 52.2° are indexed to (1 1 1), (2 2 0) and (3 1 1) planes of pure CdS (JCPDS Card No. 10-0454). It is noticed that all the films exhibit (1 1 1) preferred orientation with a cubic structure. Post-annealing of CdS thin films leads to improve the intensity of the diffraction peaks indicating the increase in the crystallinity of the films. The  $2\theta$  value of  $(1\ 1\ 1)$  diffraction peak shift slightly towards the higher angle and the full width at half maxima (FWHM) decreased from 0.34 to 0.17° with the

increase of annealing temperature. The decrease of FWHM leads to an increase in the crystallite size of the CdS thin films.



*annealed at different temperatures.*

The interplanar spacing (d) of CdS thin films is calculated using the Bragg's equation

 $2d\sin\theta = n\lambda$  (1) where  $\theta$  is Bragg angle,  $\lambda$  is the wavelength of X-ray (= 0.154 nm) and *n* is an integer. The interplanar spacing of CdS thin films decreases from 0.3365 to 0.3342 nm with the increase of annealing temperature because of a strong interaction between the vapor atoms and the substrate and realignment of grains. The lattice constant (*a*) of CdS thin films was calculated from the formula

$$
a = d (h2 + k2 + l2)1/2
$$
 (2)

for the observed *d* values corresponding to (h k l) plane obtained by X-ray diffraction data. The lattice constant (*a*) values decreases from 0.5829 to 0.5788 nm with the increase of annealing temperature due to the shift in the angular position of the prominent peak. The crystallite size (*D*) of CdS thin films is estimated from Scherer's formula [17]. The crystallite size value increased from 25.07 to 50.17 nm due to the decrease of grain boundary leading to higher conductivity in the films. The number of crystallites per unit area (*N*) is determined by using the relation

$$
N = t/D^3 \tag{3}
$$

where *t* is the thickness of the film and *D* is crystallite size of the films. The thickness of the films measured from surface profilometer is in the range of 190-320 nm. The thickness of CdS thin films decreased with the increase of annealing temperature from 150 to 300°C. The number of crystallites of CdS thin films decreases from 6.35 x  $10^{15}$  to 1.58 x  $10^{15}$  $m<sup>-2</sup>$  with the increase of annealing temperature from 150 to  $300\text{ °C}$ . The structural parameters of CdS thin films postannealed at different temperatures are given in Table 1.

anneared at different temperatures							
Annea-	2θ	<b>FWHM</b>	d-	Lattice	Crystallite	No. of	
ling	(°)	(°)	spacing	constant,	size,	crystalli	
Temp.			(nm)	a (nm)	D(nm)	tes, $N \times$	
(°C)						$10^{15}$ (m <sup>-</sup>	
<b>RT</b>	26.46	0.34	0.3365	0.5829	25.07	12.68	
150	26.55	0.27	0.3354	0.5809	31.58	6.35	
200	26.56	0.24	0.3352	0.5806	35.53	4.46	
250	26.61	0.18	0.3348	0.5798	47.38	1.88	
300	26.65	0.17	0.3342	0.5788	50.17	1.58	

Table 1. Structural parameters of CdS thin films postannealed at different temperatures

# **3.2 AFM analysis**

AFM images of (scan area: 2  $\mu$ m × 2  $\mu$ m) CdS thin films prepared at room temperature and post-annealed from 150 to 300 °C are shown in Figure 2.



*Figure 2. AFM images of CdS thin films post-annealed at different temperatures.*

From AFM images, it is observed that the grain size increases with the increase of annealing temperature which is due to the increase in grain boundary movement and mobility of the thin films. The average roughness  $(R_a)$  of the CdS thin films increased from 11.4 to 17 nm, whereas the RMS roughness  $(R_q)$  values increased from 7.7 to 13 nm with the increase of annealing temperature and reach the maximum value at 300 °C. The increase in surface roughness of the films decreases the reflective loss which consequently increases the quantum efficiency of a solar cell [18]. The maximum peak height  $(R_p)$  and peak valley  $(R_z)$  values increased with increasing annealing temperature due to the enhancement in crystallinity of the films. The negative values of skewness  $(R_{sk})$  indicate that the film had a rough surface with a low density of dips or valleys and flat regions.

When kurtosis  $(R_{ku})$  value is equal to 3, it represents the surface with a Gaussian height distribution and the surface is called Mesokurtic, but if  $R_{ku}$  < 3 the surface is flat and called Platykurtic. If  $R_{ku} > 3$ , the surface has more number of peaks than valleys. The CdS thin films exhibit the value of  $R_{ku} > 3$ and the distribution will have high number of peaks and low valleys with a spiky surface on the films. The statistical analysis of CdS thin films obtained from AFM is given in Table 2.

Annealing temp. $(^{\circ}C)$	$\mathbf{R}_{\rm a}$ (nm)	$\mathbf{R}_{\mathbf{q}}$ (nm)	R, (nm)	$\mathbf{R}_{\mathbf{p}}$ (nm)	$\mathbf{R}_{\mathbf{v}}$ (nm)	$R_{sk}$	$R_{kn}$
RT	11.4	7.7	4.4	30.1	$-31.8$	$-0.55$	4.17
150	13.5	8.5	8.3	33.6	$-38.4$	$-0.30$	3.98
200	10.2	9.3	17.2	34.1	$-26.4$	$-1.06$	5.19
250	16.2	12.8	20.6	36.7	$-39.6$	$-0.17$	5.18
300	17	13	22.1	45.3	$-43.6$	$-0.40$	4.10

Table 2. Surface roughness parameters of CdS thin films.

#### **3.3Electrical properties**

The electrical resistivity (*ρ*) of the CdS thin films were measured by using a four-probe method. The electrical resistivity of CdS thin film at room temperature is found to be  $1.47 \times 10^5$  Ω.cm, whereas its resistivity decreased from  $0.74 \times 10^4$  to  $0.12 \times 10^4$  Ω.cm with the increase of annealing temperature from 150 to 300 °C. The electrical resistivity of the films is decreased due to the decrease in density grain boundary intercrystallites and defects such as pinholes, voids, etc. The obtained electrical resistivity values are in good agreement with previously reported works [19,20]. The carrier concentration  $(n_c)$  and mobility  $(\mu_H)$  of the CdS thin films is measured from Hall effect apparatus at room temperature in a magnetic field of 0.35T [21]. The sign of Hall voltage across the samples is negative, indicating n-type semiconducting behavior of the films. The carrier concentration of CdS thin films increased from  $0.76 \times 10^{13}$  to  $1.96 \times 10^{13}$  cm<sup>-3</sup> with the increase of annealing temperature from 150 to 300 °C. The increase in the carrier concentration of the films causes shrinkage in the gap known as the bandgap narrowing, which signifies that the density of dislocations and density of grain boundaries will be decreased [22]. The carrier concentration is in the order of  $10^{13}$  cm<sup>-3</sup> for all the films and these values are in good agreement with chemical bath deposited CdS thin films [23,24]. The Hall mobility of the films increased from 110 to  $258 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$  with the increase of annealing temperature from 150 to 300 °C due to the increase in the grain size and reducing grain boundary scattering. The mean free path (*L*) of an electron can be calculated using the equation [25]

$$
L = (3\pi^2)^{1/3} (h/2\pi e^2). \; 1/\rho. \; n_c^{2/3} \tag{4}
$$

here *h* is Planck's constant, *e* is the electron charge, *ρ* is the resistivity, and  $n_c$  is the carrier concentration. The mean free path (*L*) of an electron of CdS thin films is found to be increased with the increase of annealing temperature. The longest mean free path of 4.73 nm is obtained for the film annealed at 300  $^{\circ}$ C. These values of mean free path are much shorter than the grain size measured according to XRD analysis. The effect of scattering at grain boundaries and dislocations may be neglected because the mean free path is smaller than the grain size, and scattering of conduction electrons mainly depend on ionized and neutral impurities. The electrical properties of CdS thin films post-annealed at different temperatures are given in Table 3.

Table 3. Electrical properties of CdS thin films post-annealed at different temperatures

Ann. Temp. $({}^{\circ}C)$	<b>Type</b>	Electrical resistivity, $\rho$ ( $\Omega$ . cm)	<b>Carrier</b> concentration, $\mathbf{n}_{\rm c}\times\mathbf{10}^{13}$ (cm	Hall- mobility, $\mu_{\rm H}$ $\overline{\text{cm}^2 \text{V}}$ $\text{S}^{-1}$	<b>Mean</b> free path, L (nm)
RT	n	$1.47 \times 10^5$	0.68	62	3.34
150	n	$0.74 \times 10^4$	0.76	110	3.46
200	n	$0.47 \times 10^4$	0.86	154	3.61
250	n	$0.26 \times 10^4$	1.01	231	3.80
300	n	$0.12 \times 10^4$	1.96	258	4.73

# **3.4 Optical properties**

## **3.4.1 Optical transmittance and energy band gap**

The optical transmittance spectra of the CdS thin films postannealed at different temperatures are shown in Figure 3(a). The shift in absorption edge towards longer wavelengths with an increase in post-annealing temperature indicates the growth of the particles. The average optical transmittance of CdS thin films (in the UV-Vis region) increased from 65 to 82% with the increase of annealing temperature which may be caused by lattice imperfections, reduction in voids, film thickness and increase in the crystallite size. The optical constants of CdS thin films were calculated from the equations as reported in the earlier literature [26]. The absorption coefficient (*α*) values of CdS thin films at the wavelength 550 nm increases from  $1.30 \times 10^4$  to  $2.03 \times 10^4$  $cm<sup>-1</sup>$  with the increase of annealing temperature from 150 to 300 °C. The extinction coefficient (*k*) values (at *λ*=550 nm) increases from 0.062 to 0.097 with the increase of annealing temperature from 150 to 300 °C. The optical band gap of the CdS thin films was determined by using the Tauc's formula [27,28]. Figure 3(b) shows the extrapolation of  $(ahv)^2$  versus *hν* with the intercept of the straight line on *hν* axis.



*Figure 3. (a) Optical transmittance spectra and (b) Tauc's plot for CdS thin films.*

The optical band gap of CdS thin films at room temperature is found to be 2.42 eV. The optical band gap values of the films decreased from 2.35 to 2.21 eV with the increase of post-annealing temperature from 150 to 300 °C because of increase in the particle size and quantum size effect. During annealing, the recrystallization process densified the film and eliminates the defects present in the material [29]. The obtained direct band gap values for cubic CdS thin films are in good agreement with the previously reported values [30,31]. The refractive index (*n*) of CdS thin films annealed at different temperatures was calculated using the equations as reported in the earlier literature [32]. The variation of refractive index with wavelength for CdS thin films postannealed at different temperature are shown in Figure 4(a). The refractive index (*n*) of the films decreased from 2.30 to 1.90 with the increase of annealing temperature from 150 to 300 °C due to the improvement in crystallinity of the films. H. Metin et al [33] reported similar behavior in the change of the refractive index for the CdS thin films prepared by the chemical deposition method.

## **3.4.2 Optical dispersion, group and phase velocity**

The optical dispersion study is an important parameter to understand the optical properties of the films. Figures 4(b)- (d) shows the optical dispersion, group and phase velocity of films with photon wavelength respectively. It is observed that optical dispersion *dn/dλ* of the film depends on both annealing temperature and wavelength. The group velocity,  $U<sub>g</sub>$  which describes pulse propagation through an isotropic medium with undistorted shape is given by the following relation [34]

$$
U_g = d\omega/dk = c / [n - \lambda (dn/d\lambda)] \qquad (5)
$$

where  $c$  is the velocity of light in vacuum,  $k$  is the wave number, *ω* is the angular frequency of the incident light. The wavelength dependence of the group velocity and phase velocity for the CdS thin film is shown in Figure 4(b).



*Figure 4. Variation of (a) refractive index, n ; (b) optical dispersion, dn/dλ; (c) group velocity, Ug/c and (d) Phase velocity, νp/c of CdS thin films as a function of wavelength.*

The results show that the group velocity and phase velocity for CdS thin films increases with wavelength and decreases with annealing temperature which is due to the optical dispersion in the material.

#### **3.4.3 Magneto-optical constant: Verdet coefficient V**

The Verdet coefficient (*V*) is defined as the single pass volume Faraday rotation per unit magnetic field and unit thickness. The Verdet coefficient is measured using two Glan-Taylor polarizers, but in the present study, we estimated *V* from the refractive index dispersion data [35]. The refractive index (*n*) is related to *V* with the following equation

$$
V(\lambda) = \mu_o.r.(e/2mc) \lambda \, dn/d\lambda \tag{6}
$$

where  $\mu_0$  is the vacuum permeability, *r* is the magneto-optical anomaly factor whose value is close to 0.28 for materials exhibiting covalent bond character and 1 for materials showing ionic type bond, *e* is the charge of electron, *m* is the mass of electron, *c* is the velocity of light and *dn/dλ* is the optical dispersion of the material. The wavelength dependence of the Verdet coefficient is shown in Figure 5, we noticed that the Verdet coefficient takes negative and positive values and mainly dependent on the band gap in the negative part.<br> $\overline{a}$  1.5x10



*Figure 5. Variation of Verdet coefficient dispersion of CdS thin films.*

#### **3.5 Optical dispersion parameters**

The refractive index dispersion data for CdS thin films is analyzed in terms of the single effective oscillator model proposed by Wemple-DiDomenico (WDD) [36]. The energy parameters single-oscillator energy (*Eo*) and dispersion energy  $(E_d)$  is introduced and the refractive index  $(n)$  at any photon energy *hν* is expressed by the Wemple-DiDomenico relationship [37]

$$
n^2 - 1 = 1 - E_o E_d / (E_o^2 - E^2)
$$
 (7)

The dispersion parameters  $E<sub>o</sub>$  and  $E<sub>d</sub>$  are obtained from the slope  $(E_o \tilde{E}_d)^{-1}$  and the intercept  $E_o \ell_{d}$  on the vertical axis of the straight line portion of the  $(n^2-1)^{-1}$  against  $(hv)^2$  plot as shown in Figure 6(a). The values of  $E<sub>o</sub>$  and  $E<sub>d</sub>$  obtained from the linear fitting are reported in Table 4. The values of *E<sup>o</sup>* and  $E_d$  decreases with increasing post-annealing temperature, which confirms the decrease in the optical band gap of the films with increasing annealing temperature. The singleoscillator parameters of the oscillator energy *Eo*, and the dispersion energy  $E_d$  are given by

$$
E_o^2 = M_{.1}/M_{.3}
$$
 (8)  

$$
E_d^2 = M_{.1}^3/M_{.3}
$$
 (9)

where *M−1* and *M−3* are the −1 and −3 moments of the optical spectrum, respectively [38,39]. The calculated values of *M−1* and *M−3* are given in Table 4. *M−1* values decreased from 1.31 to 0.77 and *M−3* values decreased from 0.23 to 0.18  $(eV)^2$ .

The average interband oscillator wavelength (*λo*) and average oscillator strength (*So*) for CdS thin films can be determined by using the single Sellmeier oscillator relation [40]

$$
n^2 - I = S_o \lambda_o^2 / I - (\lambda_o / \lambda)^2 \tag{10}
$$

where  $S$ <sup>*o*</sup> is the average oscillator strength, and  $\lambda$ <sup>*o*</sup> is the average oscillator wavelength. The  $S$ <sup> $o$ </sup> and  $\lambda$ <sup> $o$ </sup> are obtained from the slope and the intercept on the vertical axis of the straight-line portion of the  $(n^2 - 1)^{-1}$  versus  $\lambda^{-2}$  plot shown in Figure 6(b). The intersection with  $(n^2-1)^{-1}$  axis is  $(n_\infty^2-1)^{-1}$ and hence,  $n_{\infty}^{2}$  at  $\lambda_{o}$  equal to high frequency dielectric constant (*ε∞*).

The dielectric constant is partially due to free electrons and bound electrons as represented with the following relation [41]

$$
\varepsilon_1 = n^2 = \varepsilon_L - (e^2 N/4\pi \varepsilon_o c^2 m^*) \lambda^2 \tag{11}
$$

where  $\varepsilon_L$  is the lattice dielectric constant and  $N/m^*$  is the ratio of carrier concentration to its effective mass. The dependence of  $n^2$  with  $\lambda^2$  for CdS thin films is shown in Figure 6(c) which exhibits linear, *ε<sup>L</sup>* and *N/m\** values are estimated from the intercept and slope of the plot. It is found that the values of *ε<sup>L</sup>* and *N/m\** are increased with increasing annealing temperature, which is due to the increase in free charge carrier density and low lattice phonon interaction. The values of ε∞, *εL,* and *N/m\** are given in Table 4.

The characteristic frequency at which the material changes from metallic to dielectric response is known as plasma frequency  $(\omega_p)$ . According to the Drude free-electron model, the plasma frequency  $\omega_p$  is given as

$$
\omega_p^2 = e^2 N/m \ast \varepsilon_o \tag{12}
$$

The plasma resonance frequency  $(\omega_p)$  of all valence electrons involved in the optical transitions was calculated using the above relation and the values for CdS thin films are reported in Table 4. The increase in free carrier concentration of CdS thin films upon annealing favours the increase in plasma frequency of CdS thin films.



 $(c)$   $n^2$  *Vs*  $\lambda^2$  *for CdS thin films* 

# **3.6 Nonlinear optical properties**

Nonlinear optics study plays a key role in understanding fundamental applications of all optical circuits, optical signals processing units, and switching devices. In nonlinear media the dielectric polarization, *P* shows nonlinear behaviour with the electric field, *E* of the incident light. Thus the non-linear electron polarizability  $P_{NL}$  can be shown by the equation as given below [42,43]

$$
P = \chi^{(1)} E + P_{NL} \tag{13}
$$

where  $P_{NL} = \chi^{(2)} E^2 + \chi^{(3)} E^3$ , where  $\chi^{(1)}$  is linear optical susceptibility,  $\chi^{(2)}$  and  $\chi^{(3)}$  are second and third order nonlinear optical susceptibilities. The linear optical susceptibility,  $\chi^{(1)}$  of a medium can be written by the following equation [44]

$$
\chi^{(1)} = (n^2 - 1)/4\pi \tag{14}
$$

 $\chi^{(3)}$  is a parameter which indicates that the films are appropriate for optical switching and photonic applications. According to Miller's generalized rule, the relation between third-order nonlinear optical susceptibility,  $\chi^{(3)}$  and linear optical susceptibility,  $\chi^{(1)}$  is given as *χ*

$$
\chi^{(3)} = A \, \tilde{(\chi^{(1)})^4} \tag{15}
$$

(16)

From Eq.  $(14)$  and Eq.  $(15)$ , we can obtain the following relation

$$
\chi^{(3)} = A(n^2 - 1)^4 / (4\pi)^4 \tag{1}
$$

where  $A \approx 1.7 \times 10^{-10}$  esu, is a quantity that is independent of frequency and mostly it is the same for all materials [45,46]. The nonlinear refractive index  $n^{(2)}$  is an important parameter for manufacturing efficient and reliable optoelectronic devices. The nonlinear refractive index  $n^{(2)}$  is directly related to third-order nonlinear optical susceptibility,  $\chi^{(3)}$ . Tichy and Ticha relation gives a combination of the Miller's generalized rule and  $n_0$  obtained from WDD model which is given by the equation

$$
n^{(2)} = 12 \pi \chi^{(3)}/n_o \tag{17}
$$

Figures 7(a-c) shows the results of linear optical susceptibility  $\chi^{(1)}$ , third order non-linear optical susceptibility  $\chi^{(3)}$  and nonlinear refractive index  $n^{(2)}$  as a function of photon energy (*hν*) for CdS thin films.



*Figiure 7. Variation of (a) first order linear susceptibility, χ (1); (b) third order nonlinearsusceptibility, χ (3); and (c) nonlinear refractive index,*  $n^{(2)}$  *with energy for CdS thin films.*

The behavior of the nonlinear refractive index is same as a third order nonlinear optical susceptibility as a function of energy. It can be observed that the annealing temperature improves the non-linear response. In the present work, the third order nonlinear optical susceptibility value was observed to be high as compared to the earlier reported values for CdS thin films [47,48]. The results show that this CdS thin films exhibit large third-order nonlinear susceptibility which makes it potentially useful for nonlinear optical devices.

Table 4. Optical dispersion parameters of CdS thin films

<b>Parameter</b>	RT	150 $\rm ^{\circ}C$	200 $\rm ^{\circ}C$	250 $\rm ^{\circ}C$	300 $\rm ^{\circ}C$
$E_g$ (eV)	2.42	2.35	2.28	2.25	2.21
$E_0$ (eV)	2.96	2.39	2.17	1.78	1.77
$E_d$ (eV)	3.14	2.69	2.15	1.83	1.36
$M_{-1}$	1.31	1.03	0.99	0.91	0.77
$M_{-3} (eV^{-2})$	0.32	0.24	0.23	0.21	0.18
$\lambda_{o}(nm)$	418	520	570	696	700
$-2$ m	6.06	4.18	3.04	2.12	1.57



#### **V. CONCLUSION AND FUTURE SCOPE**

CdS thin films were successfully synthesized by sol-gel spin coating method and post-annealed in the vacuum from 150 to 300 °C for 1 hour. X-ray diffraction studies of CdS thin films show cubic structure with preferential growth along the (111) plane. The surface morphology of the CdS thin films shows that the grain size increased with the increase of annealing temperature from 150 to 300 °C. The optical transmittance and band gap of CdS thin films decreased from 2.42 to 2.21eV with the increase of annealing temperature from 150 to 300 ° due to the quantum confinement effect. The Verdet coefficient for the CdS thin films was calculated from the dispersion of the refractive index. It exhibits a spread negative range which was dependent on the optical band gap energy. The dispersion of the refractive index of CdS thin films was analyzed by applying the Wemple-DiDomenico single oscillator model. The optical dispersion parameters such as the oscillator resonance energy *Eo* and oscillator dispersion energy *E<sup>d</sup>* decreased with increasing annealing temperature. The nonlinear optical parameters linear optical susceptibility  $\chi^{(1)}$ , nonlinear optical susceptibility  $\chi^{(3)}$ , and nonlinear refractive index  $n^{(2)}$  are also reported. The future scope of CdS thin films was found to be appropriate for designing nonlinear optical devices/optoelectronic devices.

## **ACKNOWLEDGMENT**

The authors would like to convey a special gratitude to CeNSE (Center of Nano Science and Engineering), Indian Institute of Science (IISc), Bangalore, India for providing the necessary facilities for this research work under INUP (Indian Nanoelectronics Users Program).

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